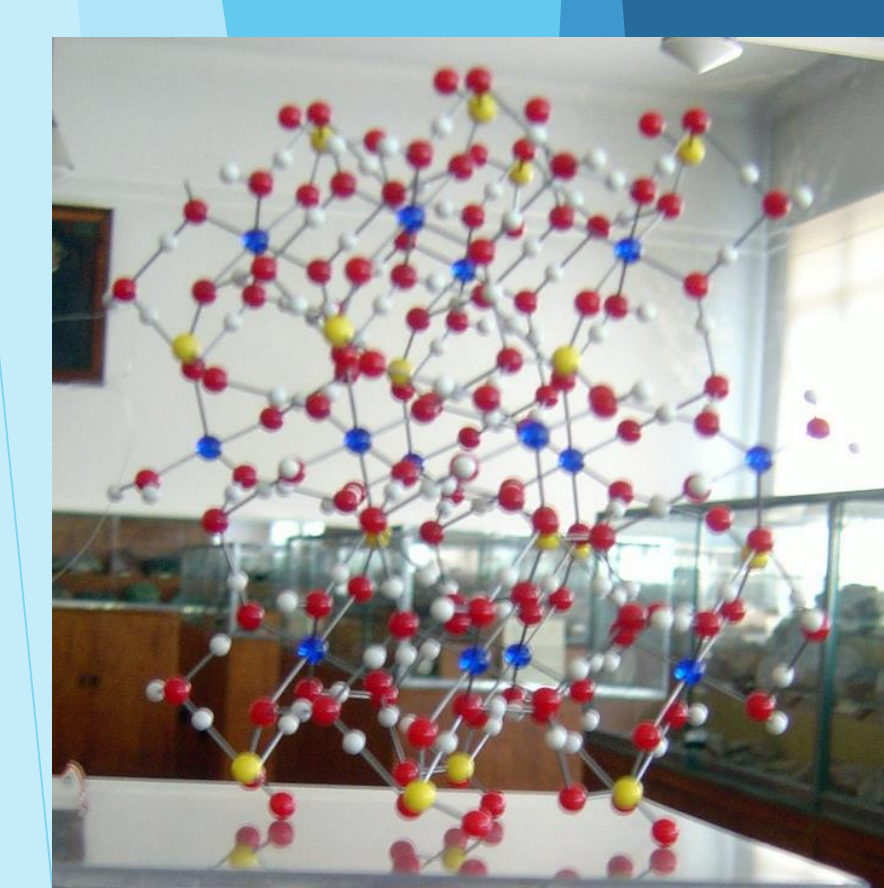
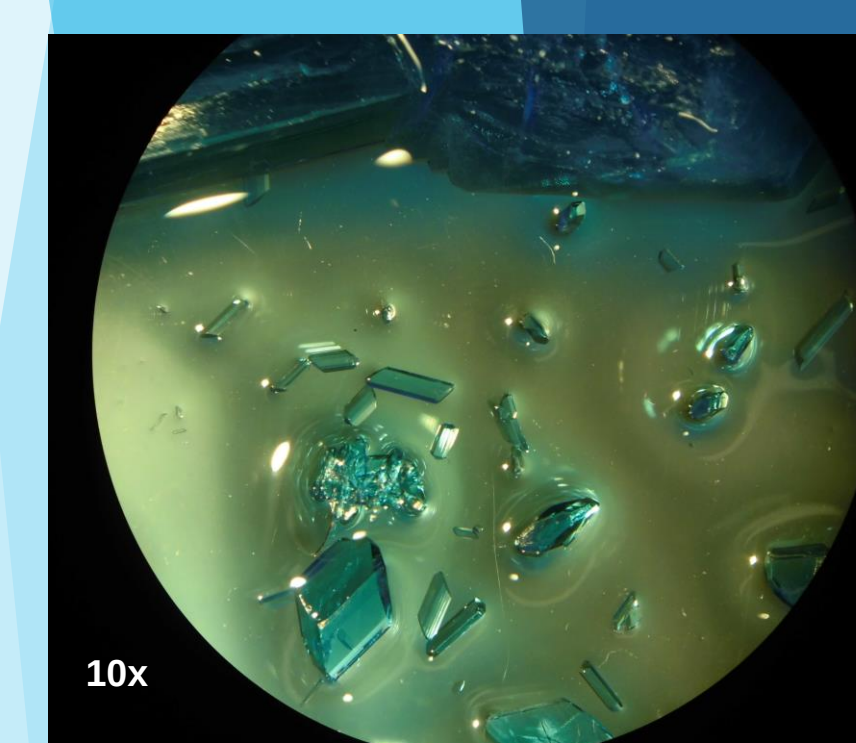
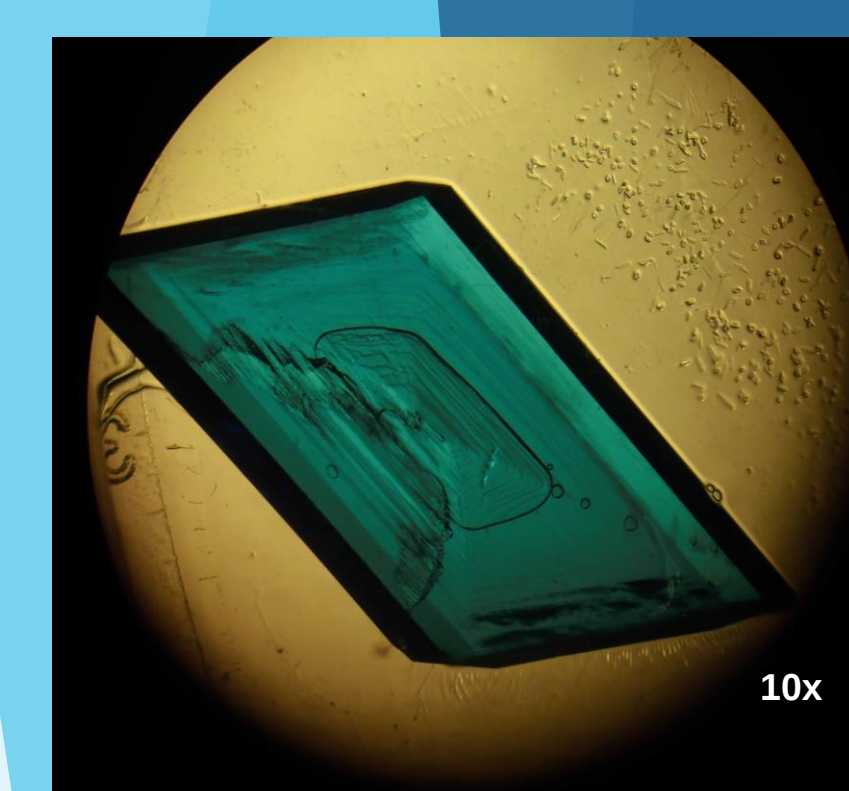


Copper (II) Sulphate
Pentahydrate $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$



50-56 g/100mL water. Any host. Best host: granite.

Growth by H-Bonding !!!!! ?



Beevers Miniature Models,
 $\times 10^8$ Crystals Lattices

Copper-Ammonium Schönite
 $\text{CuSO}_4 \cdot (\text{NH}_4)_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$

Any Host. Prepared by addition of $(\text{NH}_4)_2\text{SO}_4$ (c) to Cu^{2+} (aq), recycling solutions of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Growth enhanced by H-bonding !!!!!!!?



CRYSTAL GROWTH "ON THE ROCKS". CHEMICAL MICROSCOPY. ACTIVITIES FOR THE INTERNATIONAL YEAR OF CRYSTALLOGRAPHY 2014

II Encontro Internacional da Casa das Ciências, Porto, 14 a 16 de Julho de 2014, Instituto Superior de Engenharia do Porto

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<https://www.facebook.com/pages/Prof.ªClementina Teixeira da Cunha Pereira>

Are crystals growing better "On the Rocks"? Always? Is Crystal Growth "On the rocks" enhanced by H-Bonding with the host?

Potassium Ferricyanide, $\text{K}_3[\text{Fe}(\text{CN})_6]$
Potassium Hexacyanoferrate(III)

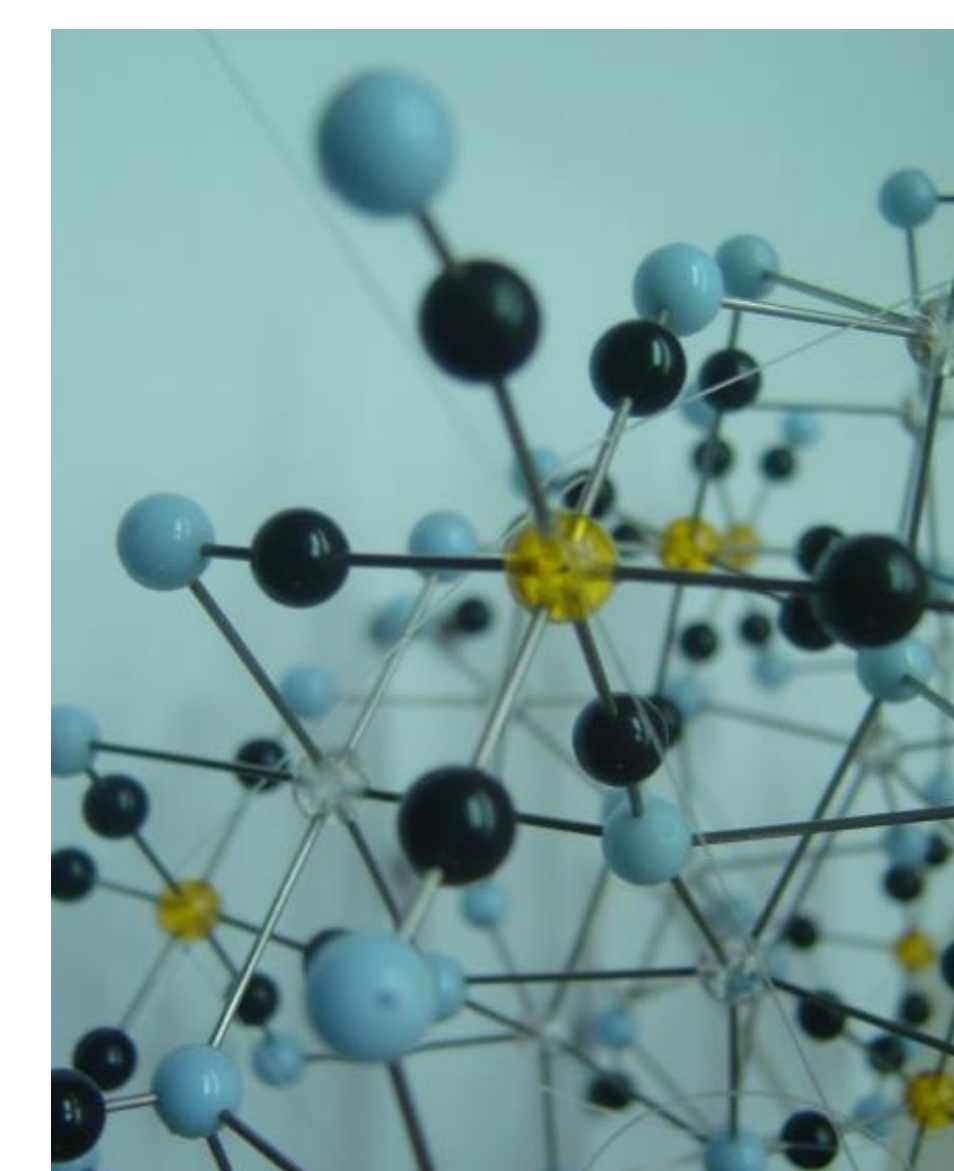
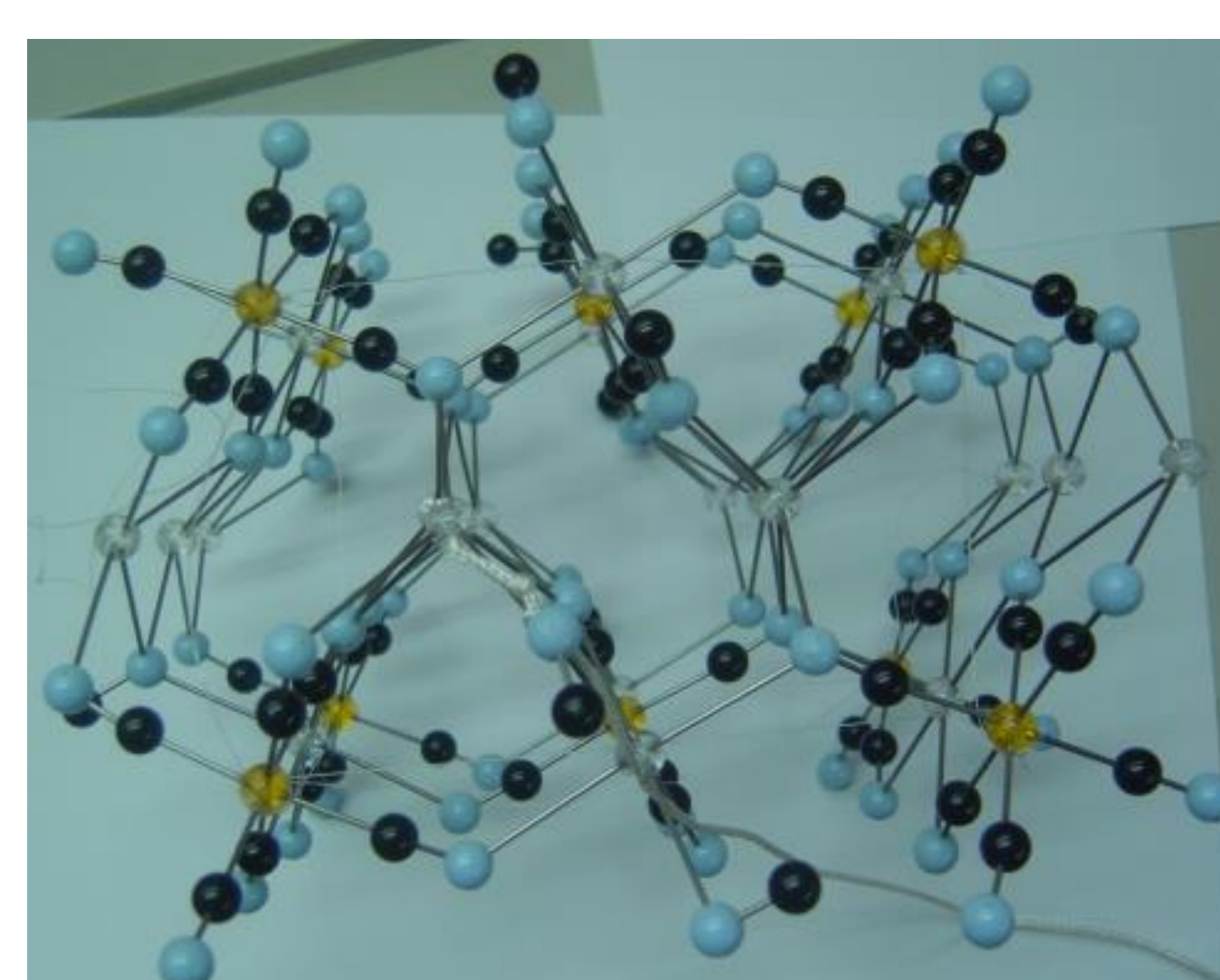
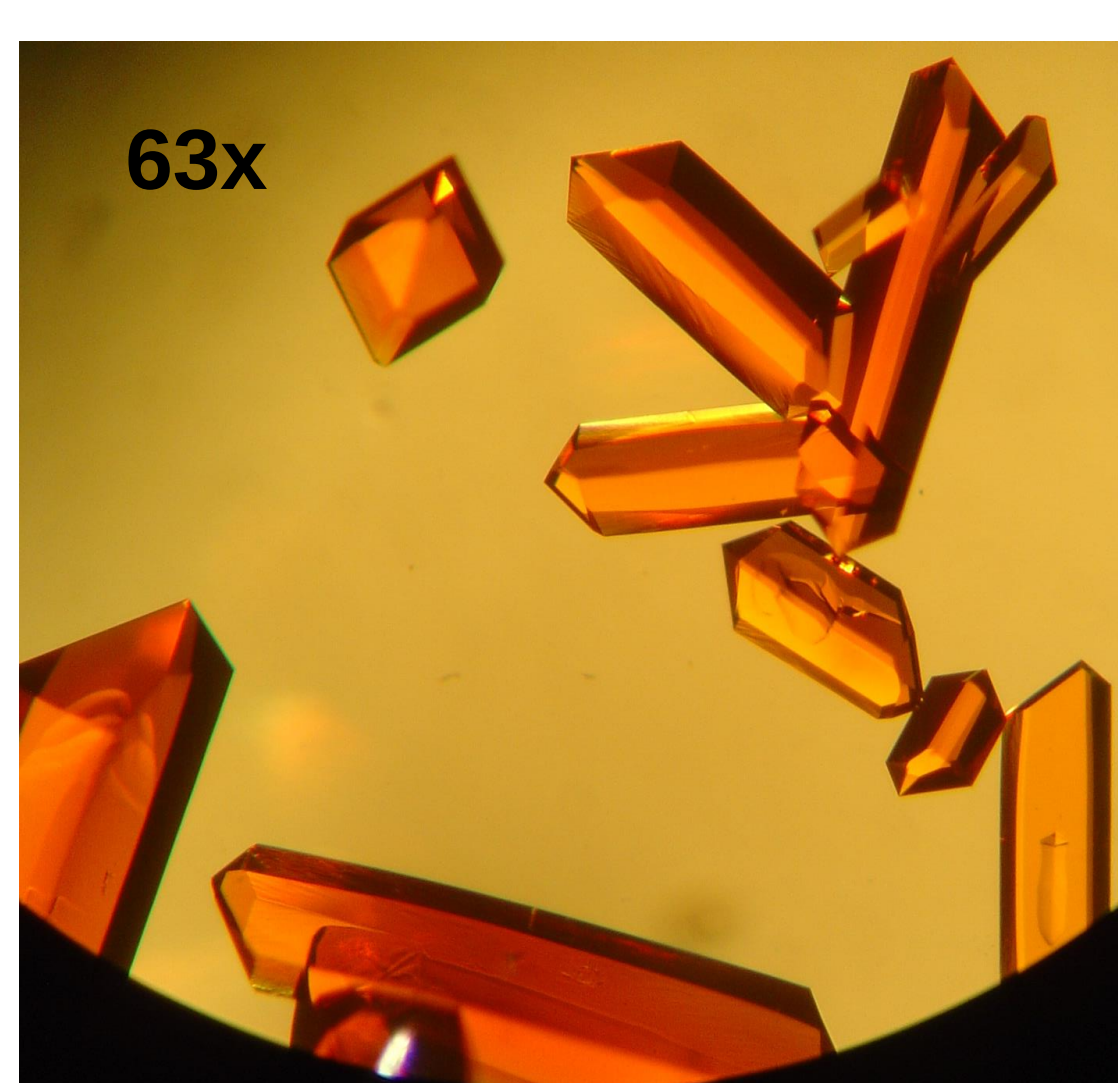
46-50 g/100 mL water. Any Host.
No enhanced growth "On the Rocks"!!!!
No hydrogen bonding!!!!!!
Crystals are growing better by homogeneous nucleation, without the host

Another example: NaCl, etc.!!!!!!

60 g/100 mL, shells



50 g/100 mL, no host!!!!!!
Growth is better without host!!!!



Beevers Miniature Model, $\times 10^8$ Crystal Lattice

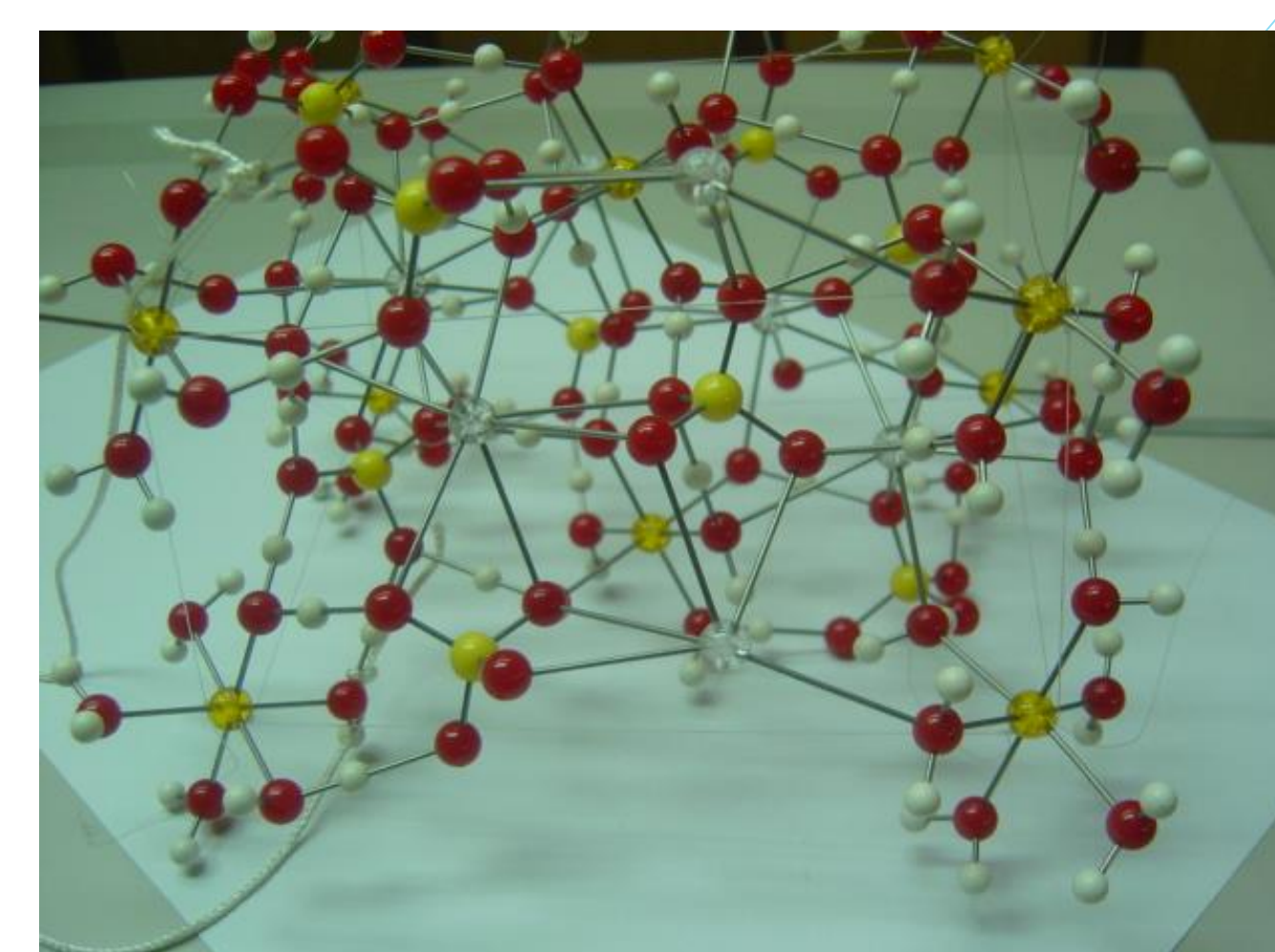
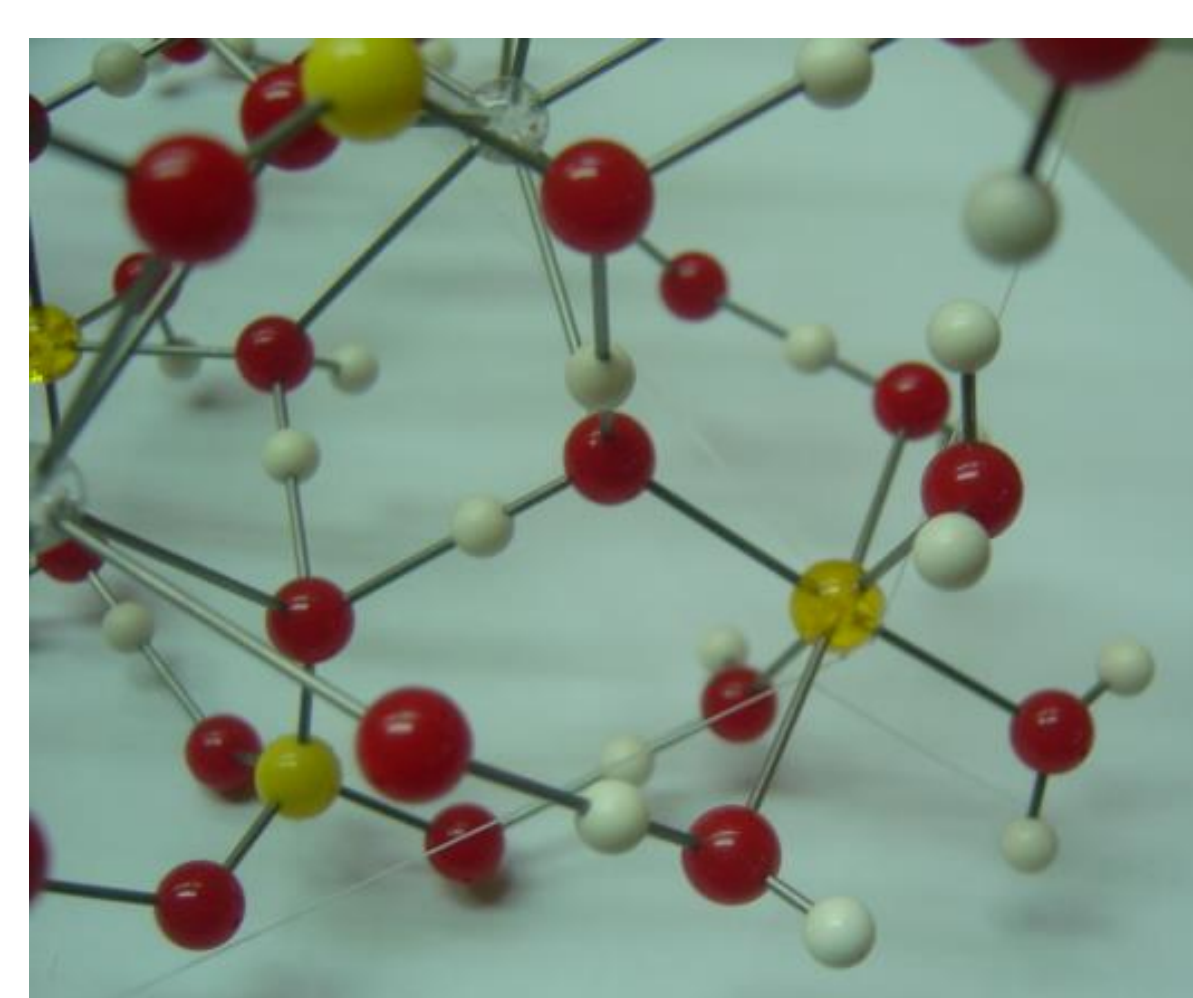
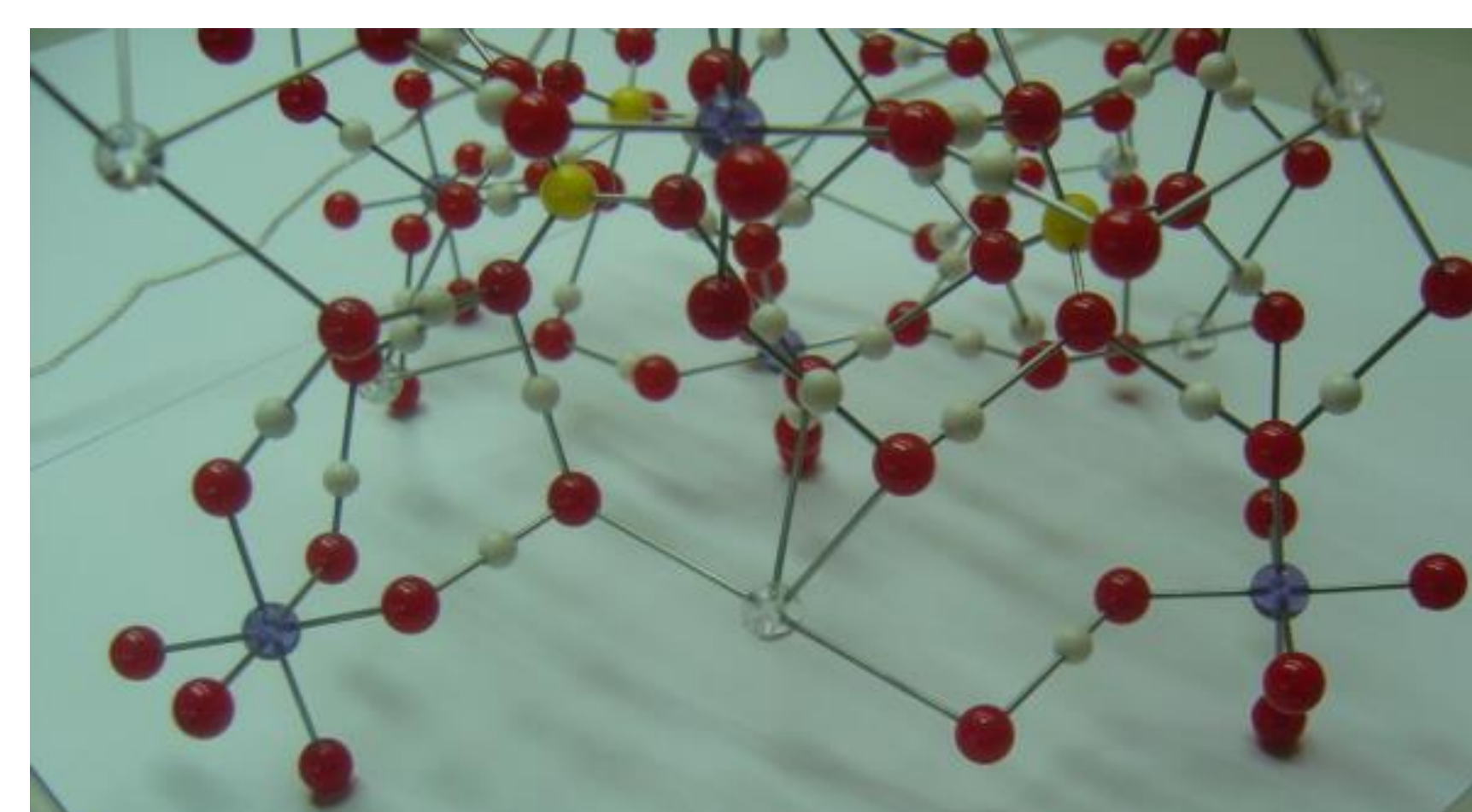
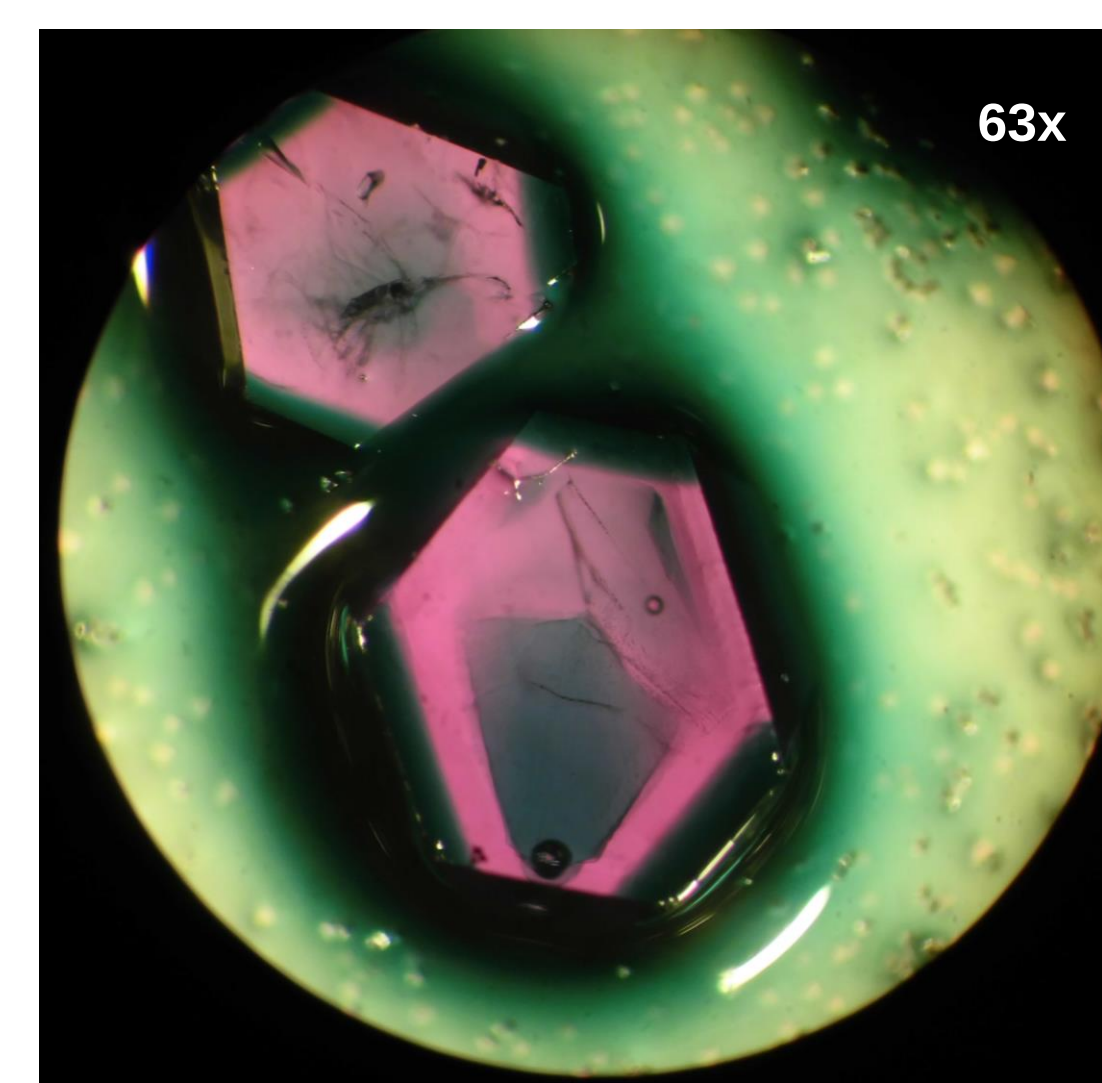
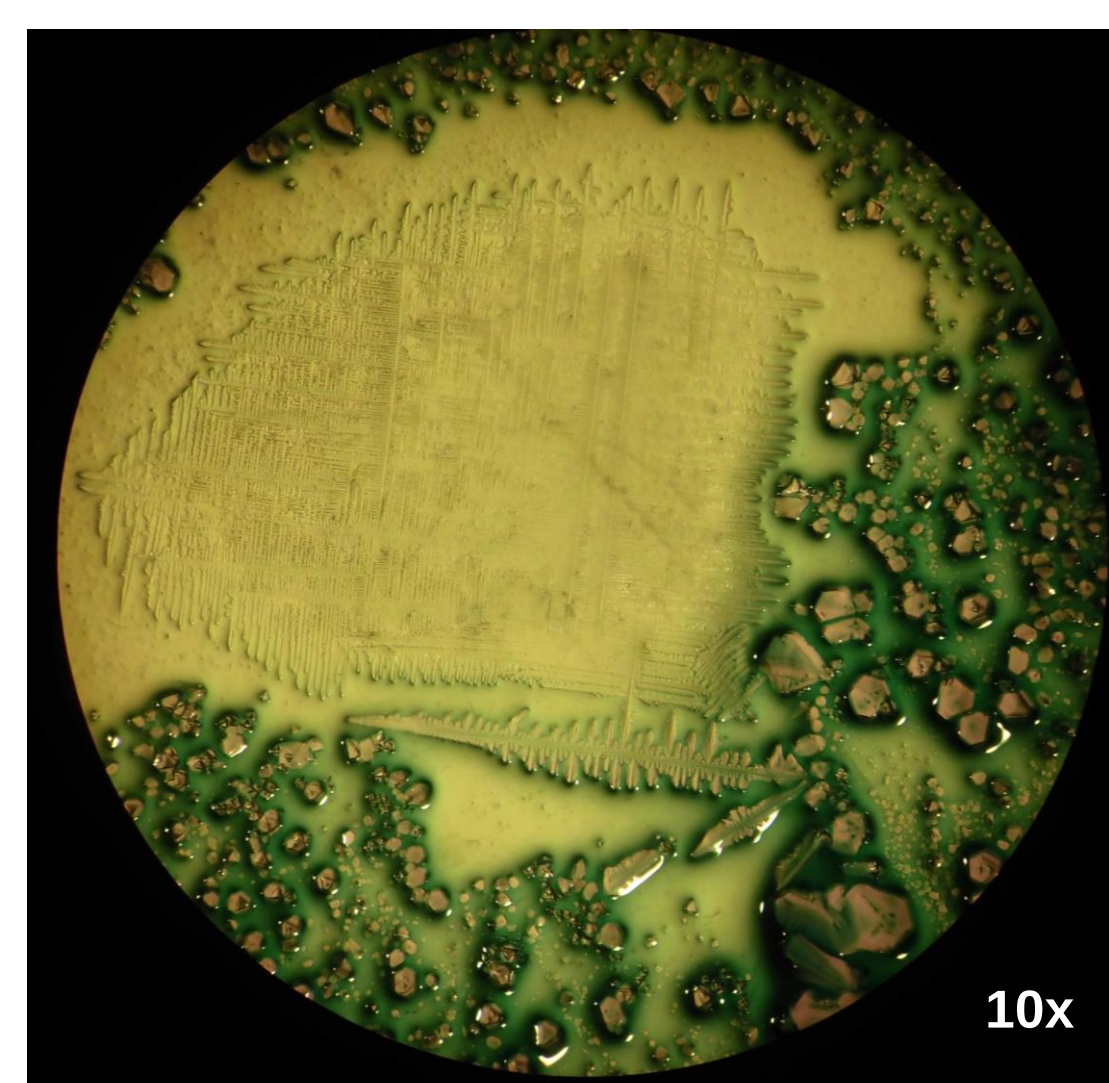
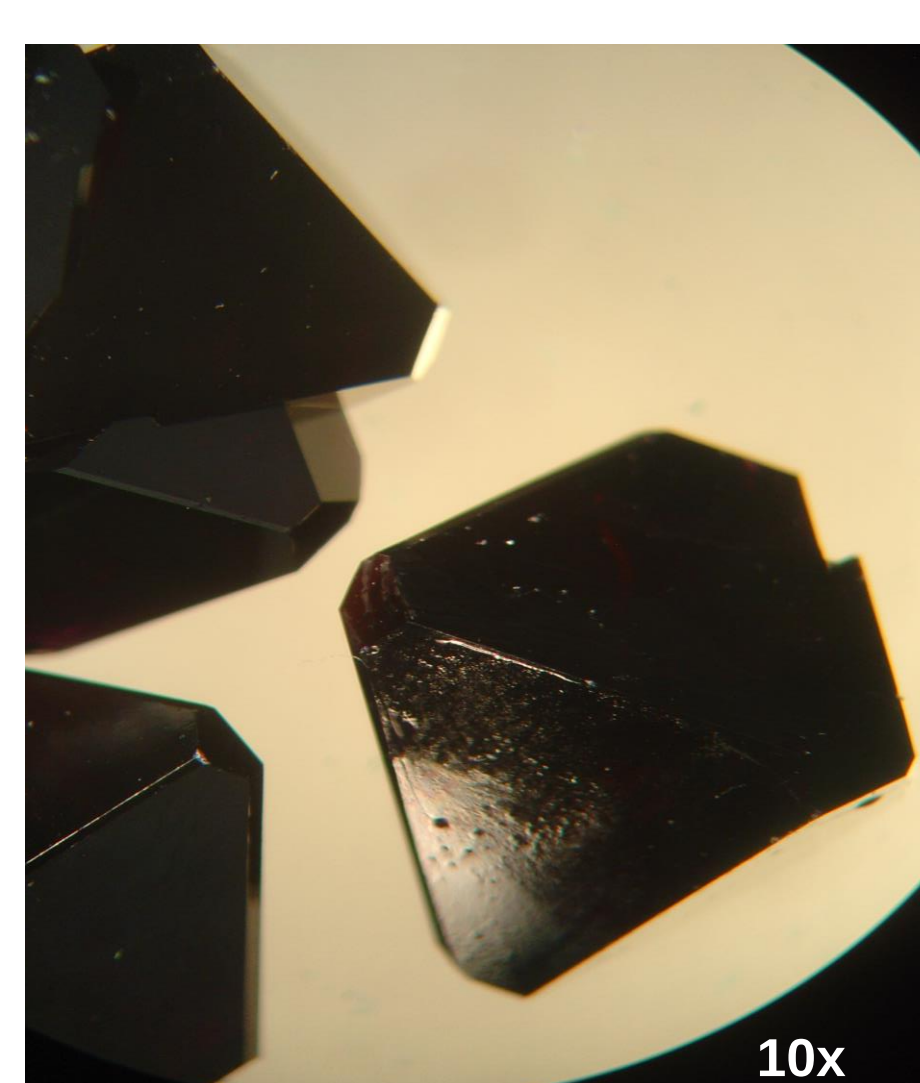


Chrome alum $\text{Cr}_2(\text{SO}_4)_3 \cdot \text{K}_2(\text{SO}_4) \cdot 24\text{H}_2\text{O}$

65g/100mL water. Volcanic rock. Any host, except low density volcanic rocks.

Growth by H-Bonding !!!!!?

Beevers Miniature Model, $\times 10^8$ Crystal Lattice Potassium Alum (similar structure)



$\text{CuSO}_4 \cdot \text{K}_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$ Beevers models, $\times 10^8$, 1 Angström=1 cm
The structure of the Cobalt-Ammonium double salt is similar, but the rate of growth is enhanced by increased H-Bonding



Cobalt-Ammonium
Schönite
 $\text{CoSO}_4 \cdot (\text{NH}_4)_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$

Any Host. Crystals prepared by the addition reaction:

70.2 g $\text{CoSO}_4 \cdot 7\text{H}_2\text{O}$

+

33 g $(\text{NH}_4)_2\text{SO}_4$ / 450 mL water.

Growth by

H-Bonding !!!!!!!?